

Highly Effective Synthetic Methods for Substituted 2-Arylbenzofurans Using [3,3]-Sigmatropic Rearrangement: Short Syntheses of Stemofuran A and Eupomatenoid 6

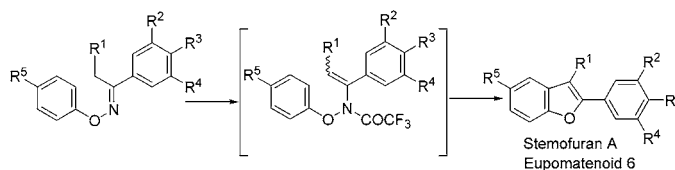
Okiko Miyata, Norihiko Takeda, and Takeaki Naito*

Kobe Pharmaceutical University, Motoyamakita, Higashinada, Kobe 658-8558, Japan

taknaito@kobepharma-u.ac.jp

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ABSTRACT



A new and efficient synthesis of 2-arylbenzofurans has been achieved via a route involving acylation and subsequent [3,3]-sigmatropic rearrangement of oxime ethers. Its synthetic utility is demonstrated by a short synthesis of stemofuran A and eupomatenoid 6 in which no procedure for protection of the phenolic hydroxyl groups is needed.

Benzo[*b*]furans carrying functional groups are attractive targets of organic synthesis because of their biological activity and their wide occurrence in nature.¹ As a consequence, there has been growing interest in developing general and versatile synthetic methods for benzofuran derivatives.² Some 2-arylbenzofurans are inhibitors of cell proliferation and platelet-activating factor, and some of them show good fungicidal, insecticidal, and cytotoxic activities.^{1j–l} Therefore, 2-arylbenzofurans have been a new subject of synthetic studies for the development of biologically active compounds.³

We have now developed a new preparative route to 2-arylbenzofurans **1** using the successive reactions of oxime ethers. The route involves acylation, [3,3]-sigmatropic rearrangement, and intramolecular cyclization reactions.⁴ Furthermore, this method was successively applied to the short synthesis of stemofuran A **2**⁵ and eupomatenoid 6 **3**,⁶ the latter of which has shown antifungal, insecticidal, and anti-oxidant activities (Figure 1).

We first investigated the substituent effects in the reaction of oxime ethers **4**, prepared by condensation of *O*-phenyl-

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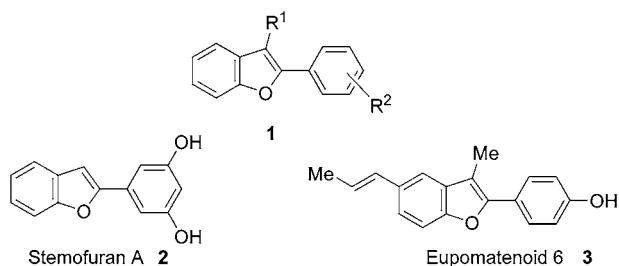
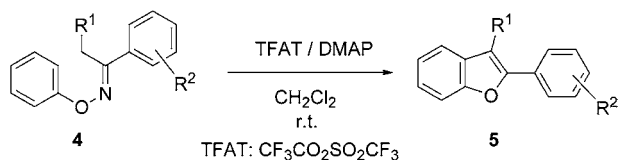


Figure 1. 2-Arylbenzofurans.

hydroxylamine and various acetophenones, with trifluoroacetyl triflate (TFAT),⁷ which is a powerful acylating agent (Table 1).⁸ When unsubstituted oxime ether **4a** was treated

Table 1. Conversion of Oxime Ethers into 2-Arylbenzofurans^a



entry	4	R ¹	R ²	time (h)	5	yield ^b (%)
1	4a	H	H	1	5a	99
2	4b	H	<i>p</i> -Br	1.5	5b	96
3	4c	H	<i>p</i> -NO ₂	5	5c	85 (9)
4	4d	H	<i>p</i> -OH	2	5d	84
5	4d	H	<i>p</i> -OH	2	5n^c	92
6	4e	H	<i>p</i> -OMe	2	5e	15 (12)
7	4f	Me	H	2	5f	82
8	4g	Me	<i>p</i> -Br	2	5g	91
9	4h	Me	<i>p</i> -OH	2	5h	86
10	4i	Me	<i>p</i> -OMe	2	5i	26 (13)
11	4j	H	<i>m</i> -Br	2	5j	94
12	4k	H	<i>m</i> -NO ₂	5	5k	95
13	4l	H	<i>m</i> -OH	2	5l	86
14	4m	H	<i>m</i> -OMe	1.5	5m	93

^a TFAT (5 equiv) and DMAP (3 equiv) were used. ^b Yields in parentheses are for the recovered starting material. ^c **5n** (R² = *p*-OCOCF₃).

with TFAT (5 equiv) in the presence of (dimethylamino)-pyridine (DMAP) (3 equiv) at room temperature, the expected 2-phenylbenzofuran **5a** was obtained in quantitative yield (entry 1). Similarly, the reaction of oxime ethers **4b** and **4c** having the *p*-bromo- and *p*-nitrophenyl groups which are electron-withdrawing group proceeded smoothly to give

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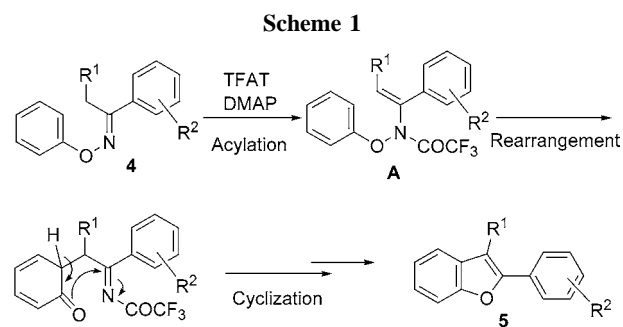
(8) When trifluoroacetic anhydride, which is the usual trifluoroacetyl agent, was used, the oxime ether could not give 2-arylbenzofuran.

2-arylbenzofurans **5b** and **5c** in good yields, respectively (entries 2 and 3). Even when a phenolic hydroxyl group is present as an electron-donating group at the para position, the desired benzofuran **5d** was obtained in 84% yield after chromatography using silica gel (entry 4). Direct recrystallization of the crude product obtained from oxime ether **4d** gave benzofuran **5n** having a *p*-trifluoroacetoxy group (entry 5). However, the oxime ether **4e** with a *p*-methoxyl group gave the benzofuran **5e** only in low yield with a small amount of starting material **4e** (entry 6).

A similar trend was observed in the reaction of oxime ethers **4f–i**, prepared from propiophenones (entries 7–10). The oxime ethers **4f–h** were subjected to reaction with TFAT to give the 2-aryl-3-methylbenzofurans **5f–h** in good to excellent yields while the oxime ether **4i** having a *p*-methoxyl group gave corresponding benzofuran **5i** in 26% yield.

The next substrate of choice was oxime ethers **4j–m** with a meta-substituted phenyl group. Under similar reaction conditions, the oxime ethers **4j–l** were converted into the desired benzofurans **5j–l** in good yield (entries 11–13). The reaction of substrate **4m** with a *m*-methoxyl group also proceeded smoothly to afford the desired benzofuran **5m** in excellent yield (entry 14).

A plausible reaction pathway based on these results is shown in Scheme 1. As shown in our previous studies,⁴ the

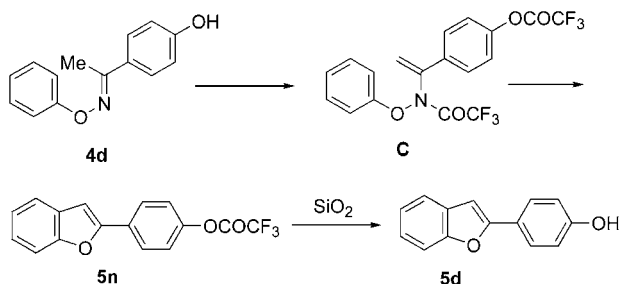


oxime ethers **4** are acylated with TFAT to form the trifluoroacetylated compounds **A**. These would immediately rearrange to give the dienones **B**, which are converted into the 2-arylbenzofurans **5** via cyclization followed by elimination of trifluoroacetamide.

In the case of oxime ether **4d** with the hydroxyl group, the acylation proceeds at both the imine part and the hydroxyl group to form diacylated enehydroxylamine **C**, which can then rearrange to afford the acylated 2-arylbenzofuran **5n**. This labile benzofuran **5n** was readily hydrolyzed during column chromatography using silica gel to give the phenolic benzofuran **5d** (Scheme 2). Similarly, oxime ethers **4h** and **4l** would give the desired 2-arylbenzofurans **5h** and **5l** via the corresponding diacylated intermediates.

As mentioned above, the series of reactions leading to the formation of 2-arylbenzofurans proceeded smoothly except for oxime ethers having a *p*-methoxyl group (**4e** and **4i**). This

Scheme 2

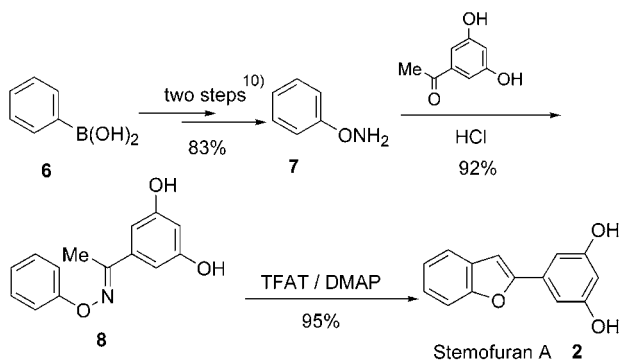


is due to the electron-donating ability of the *p*-methoxy group. The electron-donating ability lowers the reactivity of the intermediates **B** formed from **4e** and **4i** toward nucleophilic attack of the carbonyl oxygen.

The high potential of this reaction for synthesis has been proved by the practical synthesis of natural 2-arylbenzofurans. We chose stemofuran **A** **2**⁵ and eupomatenoid **6** **3**⁶ as our targets which have a hydroxyphenyl group at the 2-position of the benzofuran skeleton. We succeeded in the total synthesis of these natural products without a step for protection of the phenolic hydroxyl groups.

At first, we examined the preparation of stemofuran **A** **2**, recently isolated from *Stemona collinsae*⁵ (Scheme 3). Al-

Scheme 3

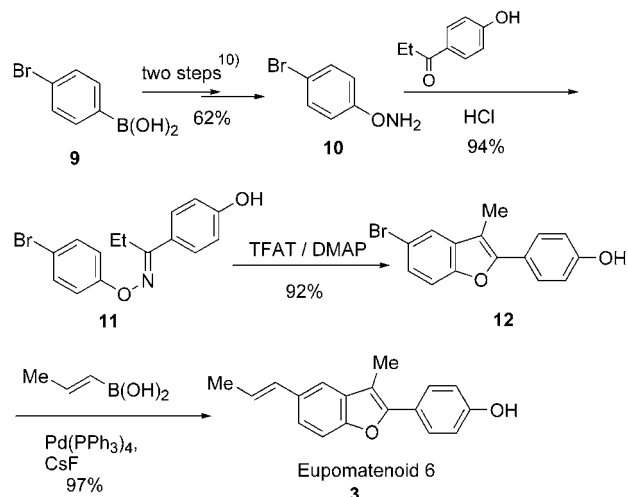


though Pasturel et al.⁹ synthesized stemofuran **A** **2** in 44% overall yield from 2-hydroxybenzaldehyde, the synthesis includes many transformations involving protection and deprotection of a hydroxyl group. *O*-Phenylhydroxylamine **7** was prepared from the phenylboronic acid **6** in two steps according to the literature procedures.¹⁰ The condensation of **7** with dihydroxyacetophenone gave the oxime ether **8**, which was subjected to our reaction conditions to give stemofuran **A** **2** in 95% yield. The spectral data of this benzofuran **2** are identical with those reported in the literature.⁵ Thus, we have succeeded in a four-step synthesis with 72% overall yield.

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Scheme 4



Another target of our synthesis is eupomatenoid **6** **3**⁶ (Scheme 4). Known syntheses of eupomatenoid **6** **3** reported by two groups^{11,12} include the protection and deprotection reactions of the hydroxyl group.

The condensation of *O*-phenylhydroxylamine **10**,¹⁰ prepared from boronic acid **9**, with propiophenone proceeded effectively to give the oxime ether **11** in excellent yield, which was then treated with TFAT to give the desired benzofuran **12** in 92% yield.

Finally, the introduction of a propenyl group was achieved by using Suzuki coupling with propenyl boronic acid to give eupomatenoid **6** **3** in 52% overall yield from **9** in five steps. Physical and spectral properties of **3** were identical with those of natural eupomatenoid **6** **3** reported in the literature.⁶ Our synthesis is superior to those reported by both Bach's¹² and Stevenson's¹¹ groups in both yield and number of steps.

In conclusion, we have now established a novel and potential synthetic route to 2-arylbenzofurans. The reaction proceeds via sequential acylation, rearrangement, and cyclization of oxime ethers under the mild conditions. Furthermore, we have succeeded in the short synthesis of stemofuran **A** and eupomatenoid **6**.

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Supporting Information Available: Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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